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## Sulfate-chloride exchange transport in a glioma cell line

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Transport of  $\text{SO}_4^{2-}$  was studied in the glioma cell line LRM55 to determine whether it is mediated by the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger or the  $\text{K}^+/\text{Cl}^-$  cotransporter previously described in these cells (Wolpaw, E.W. and Martin, D.L. (1984) *Brain Res.* 297, 317–327).  $^{35}\text{SO}_4^{2-}$  influx was saturable with  $\text{SO}_4^{2-}$ . External  $\text{SO}_4^{2-}$  stimulated  $^{35}\text{SO}_4^{2-}$  efflux, indicating an exchange mechanism. External  $\text{Cl}^-$  was a competitive inhibitor of  $^{35}\text{SO}_4^{2-}$  influx. Internal  $\text{Cl}^-$  stimulated  $^{35}\text{SO}_4^{2-}$  influx and external  $\text{Cl}^-$  stimulated  $^{35}\text{SO}_4^{2-}$  efflux, indicating that  $\text{Cl}^-$  is an exchange substrate for the  $\text{SO}_4^{2-}$  carrier. Also,  $\text{SO}_4^{2-}$  flux was sensitive to SITS, DIDS and furosemide. However, saturating external  $\text{SO}_4^{2-}$  did not inhibit  $^{36}\text{Cl}^-$  influx and did not inhibit  $^{36}\text{Cl}^-$  efflux via the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger. Moreover,  $\text{K}^+$  did not stimulate  $^{35}\text{SO}_4^{2-}$  influx as it does  $\text{Cl}^-$  influx. These findings indicate that  $\text{SO}_4^{2-}$  transport into these cells is mediated by an exchange carrier distinct from both the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger and the  $\text{K}^+/\text{Cl}^-$  cotransporter. While  $\text{Cl}^-$  is an alternative substrate for the  $\text{SO}_4^{2-}$  porter, this carrier is responsible for only a minor fraction of total  $\text{Cl}^-$  flux in these cells.

### Introduction

Anion transport across the astroglial cell membrane appears to be an important mechanism for maintenance of the extracellular environment in the central nervous system [1,2]. Two major electroneutral anion transport systems are present in the astroglial cell membrane: a cation/anion cotransporter for  $\text{K}^+$  and  $\text{Cl}^-$ , and an anion exchanger for  $\text{Cl}^-$  and  $\text{HCO}_3^-$  [3–5].  $\text{Cl}^-$  transport systems are also present in many other cells. Of these, the erythrocyte anion exchanger is the most

fully characterized. Under physiologic conditions it transports  $\text{Cl}^-$  and  $\text{HCO}_3^-$  and in vitro it also carries out homo- and hetero-exchange of  $\text{SO}_4^{2-}$  (reviewed in Refs. 6 and 7). Ehrlich ascites tumor cells also exchange  $\text{SO}_4^{2-}$  and  $\text{Cl}^-$  and may have additional  $\text{Cl}^-$  porters that do not carry  $\text{SO}_4^{2-}$  [8,9].

In the work presented here we sought to determine whether  $\text{SO}_4^{2-}$  transport in the glioma cell line LRM55 is mediated by either of the astrocytic  $\text{Cl}^-$  transporters previously described in these cells [5]. Our results show that  $\text{SO}_4^{2-}$  transport in LRM55 cells is carrier-mediated and that it occurs by an exchange mechanism. Although the  $\text{SO}_4^{2-}$  porter is able to transport  $\text{Cl}^-$  and is sensitive to compounds that inhibit  $\text{Cl}^-$  transport, our kinetic data indicate that this  $\text{SO}_4^{2-}$  porter is not the same as the major  $\text{Cl}^-/\text{HCO}_3^-$  exchanger or the  $\text{K}^+/\text{Cl}^-$  cotransporter previously described [5]. Rather, our findings indicate that the  $\text{SO}_4^{2-}$  porter in LRM55 cells is a separate carrier altogether.

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Abbreviations: DIDS, 4,4'-diisothiocyanostilbene-2,2'-disulfonic acid; Hepes, 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid; SITS, 4-acetamido-4'-isothiocyanostilbene-2,2'-disulfonic acid.

## Materials and Methods

### Methods

The methods used in this study were based on those previously described in detail [5,10].

**Cell culture.** LRM55 cells were originally cloned from a mixed glioma isolated from the spinal cord of a Fisher rat [10,11]. The cells were maintained in 100-mm diameter culture dishes and were seeded in 24-well culture dishes 6–9 days prior to each experiment. All experiments were carried out in 24-well dishes with cells from passages 6 to 26. At the time of the experiments, each well contained an average of 109  $\mu$ g protein.

**Flux assays.** After preincubation of the cells in Hepes-buffered Hanks' balanced salt solution (HBHS) at 37°C for at least 20 min, uptake was initiated by replacing the HBHS with 200  $\mu$ l of an osmotically equivalent salt solution containing the radioactively labeled uptake substrate. Uptake was terminated by rinsing the wells three times with 1 ml of HBHS at room temperature. All rinsing in these assays was performed using a syringe pipettor which was modified to direct the rinse solution evenly around the walls of the well and which incorporated an aspirator tube to remove the rinse solution immediately. By this method, cells could be rinsed three times in 7–10 s. Since over 95% of the extracellular fluid was removed in the first rinse and since the time required for all three rinses was very short compared to the total flux time, error due to uptake during the rinse procedure was negligible. To convert measured cpm to mol of influx substrate, we assumed that the specific activity of the labeled substrate in the medium did not change appreciably over the course of the experiment. This assumption was justified since the internal pool of  $\text{SO}_4^{2-}$  or  $\text{Cl}^-$  was no more than 0.2% of the total  $\text{SO}_4^{2-}$  or  $\text{Cl}^-$  in the experimental well.

In efflux experiments, cells were loaded with the labeled substrate by incubation in 200  $\mu$ l of the labeled compound in HBHS. After a loading period of at least 30 min, the labeled medium was replaced with 800  $\mu$ l or 1 ml of the appropriate efflux medium containing no label. Efflux was terminated by drawing off the medium and rapidly rinsing the well three times. Efflux was calculated as the percentage of the total label which remained

in the cells at given time points after changing to unlabeled medium. Efflux was plotted as a function of time on semilogarithmic axes.

**Correction for residual extracellular tracer.** Since flux measurements were always made by assaying the cellular content of radiolabeled compounds, it was necessary to correct for residual label in the extracellular space. Extracellular residual label was measured in separate wells by removing HBHS, adding 200  $\mu$ l of labeled incubation medium, and immediately rinsing three times. This amount was then subtracted from the total measured label for each flux data point. This method of measuring label in the residual extracellular fluid gave results similar to those obtained using impermeant molecules such as labeled sucrose and mannitol.

**Scintillation counting and protein analysis.** The cellular contents of the wells were dissolved and transferred to scintillation vials, radioactivity was measured by liquid scintillation and cellular protein was measured as previously described [5].

**Solutions.** The HBHS used in these experiments comprised 10 mM Hepes/11 mM glucose/130 mM  $\text{Na}^+$ /4.5 mM  $\text{K}^+$ /136 mM  $\text{Cl}^-$ /1.1 mM  $\text{Ca}^{2+}$ /0.71 mM  $\text{Mg}^{2+}$ /0.21 mM  $\text{SO}_4^{2-}$ /1.4 mM phosphate, and was adjusted to pH 7.3 with NaOH. Low- $\text{Cl}^-$  media were based on this composition, with isosmotic replacement of NaCl by sucrose. For  $\text{Cl}^-$ -free media, gluconate replaced the balance of the  $\text{Cl}^-$ . All drug-containing solutions were made up fresh each day. SITS and DIDS solutions were kept in opaque vessels and used in dim light.

In all experiments, the contributions of  $^{35}\text{SO}_4^{2-}$  and  $^{36}\text{Cl}^-$  to the total concentrations of  $\text{SO}_4^{2-}$  and  $\text{Cl}^-$  were considered in computing concentrations of these species.

### Materials

The following isotopes were obtained from New England Nuclear, Boston, MA:  $\text{Na}_2^{35}\text{SO}_4$  (in water, 10–1000 mCi/mmol),  $^{36}\text{Cl}^-$  (as HCl, 6.1–6.4 mCi/g), and L-[4,5- $^3\text{H}$ ]leucine (58 Ci/mmol). SITS and DIDS were obtained from Pierce Chemical Company, Rockford, IL. Furosemide (5-[aminosulfonyl]-4-chloro-2-[(furanylmethyl)-amino]benzoic acid) was a gift from Hoechst-Roussel Pharmaceuticals, Inc., Somerville, NJ.

## Results

### Sulfate influx

Since early experiments had shown that external  $\text{Cl}^-$  inhibited  $\text{SO}_4^{2-}$  influx, subsequent measurements of  $\text{SO}_4^{2-}$  influx were made in low- $\text{Cl}^-$  or  $\text{Cl}^-$ -free media. The time-course of  $^{35}\text{SO}_4^{2-}$  influx in 7 mM  $\text{Cl}^-$  medium over a period of 90 min is shown in Fig. 1. An initial rapid burst of uptake of the labeled substrate reached a maximum by 15 min. This was followed by a decrease to a level that remained constant from 30 to 90 min, indicating that the extracellular  $^{35}\text{SO}_4^{2-}$  had equilibrated with the intracellular pool. The unusual shape of this curve can be accounted for by the relationship between  $\text{Cl}^-$  and  $\text{SO}_4^{2-}$  transport to be described below and explained in detail in Discussion.

### Kinetics of sulfate transport

Initial velocities of  $\text{SO}_4^{2-}$  transport were measured in  $\text{Cl}^-$ -free medium over a range of  $\text{SO}_4^{2-}$  concentrations from 0.2 to 5 mM. A typical experiment is shown in Fig. 2. Initial rates could be described by the simple Michaelis-Menten equation. Kinetic parameters were obtained by fitting this equation to the data by a computer program

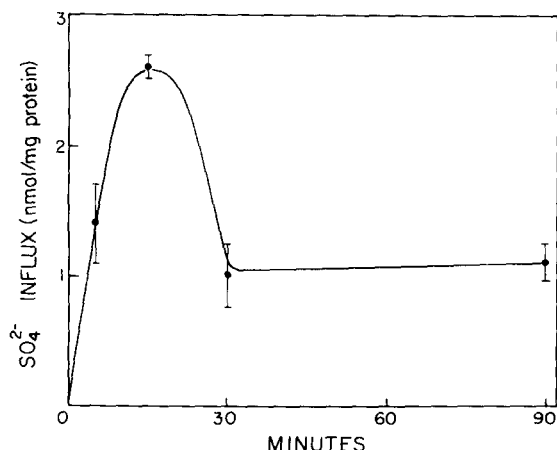


Fig. 1. Influx of  $^{35}\text{SO}_4^{2-}$  at 37°C. Influx medium (low- $\text{Cl}^-$  HBHS) contained all of the components of HBHS except that NaCl was reduced by equiosmolar replacement with sucrose so that  $[\text{Cl}^-]$  was 7 mM and  $[\text{Na}^+]$  was 9 mM. The concentration of  $\text{SO}_4^{2-}$  (including  $^{35}\text{SO}_4^{2-}$ ) was 0.21 mM. Influx was initiated by replacing the preequilibrating HBHS with the labeled medium and was terminated by rinsing three times in HBHS. Each point is the mean of 3–6 determinations ( $\pm$  S.E.).

based on the method of Eisenthal and Cornish-Bowden [12]. The apparent  $K_m$  was 0.24 mM and the apparent  $V_{\max}$  was 1.2 nmol/min per mg protein (means of two experiments). The  $V_{\max}$  for  $\text{SO}_4^{2-}$  flux is much smaller than the  $V_{\max}$  for total  $\text{Cl}^-$  flux (280 nmol/min per mg protein) in these cells [5].

### Steady-state sulfate content

The steady-state concentration of  $\text{SO}_4^{2-}$  in LRM55 cells was determined at 37°C after at least 30 min of uptake of  $^{35}\text{SO}_4^{2-}$  from HBHS containing 0.21 mM  $\text{SO}_4^{2-}$ . Under these conditions, the cellular  $\text{SO}_4^{2-}$  content was  $0.66 \pm 0.03$  nmol/mg protein (mean from nine experiments  $\pm$  S.E.). Since the intracellular space of LRM55 cells is between 2.3 and 3.9  $\mu\text{l}/\text{mg}$  protein [10,13], the internal concentration of  $\text{SO}_4^{2-}$  was between 0.17 and 0.29 mM. Thus, the internal and external concentrations of  $\text{SO}_4^{2-}$  were approximately equal.

### Efflux of sulfate and homo-exchange

To explore the possibility that  $\text{SO}_4^{2-}$  transport occurred by an exchange mechanism, we first measured the efflux of  $^{35}\text{SO}_4^{2-}$  into HBHS (Fig. 3A, lower curve). The initial rate of  $\text{SO}_4^{2-}$  efflux was 0.12 nmol/min per mg protein (mean from two experiments). After 10 min of incubation, only 40% of the label remained in the cells.

We then measured the efflux of  $^{35}\text{SO}_4^{2-}$  in the presence and absence of 10 mM external  $\text{SO}_4^{2-}$ , with no other anions (except the Hepes buffer) included in the efflux medium (Fig. 3B). In the absence of external anions,  $^{35}\text{SO}_4^{2-}$  efflux was slow, with only 8.5% of the label lost from the cells after 5 min of incubation. The presence of 10 mM external  $\text{SO}_4^{2-}$  caused a 2.8-fold increase in the initial rate of  $^{35}\text{SO}_4^{2-}$  transport out of the cells. This trans-stimulation by external  $\text{SO}_4^{2-}$  suggests that  $\text{SO}_4^{2-}$  transport is mediated by an exchange porter which can carry  $^{35}\text{SO}_4^{2-}$  from inside to outside in exchange for external  $\text{SO}_4^{2-}$ .

### Effect of chloride on sulfate flux

The effect of  $\text{Cl}^-$  on  $\text{SO}_4^{2-}$  flux was also examined. As external  $[\text{Cl}^-]$  was varied from 136 to 7.3 mM,  $\text{SO}_4^{2-}$  influx increased from  $0.09 \pm 0.002$  to  $0.43 \pm 0.02$  nmol/min per mg protein (means  $\pm$  S.E.;  $n = 3$ ). Measurements of  $\text{SO}_4^{2-}$  influx as a

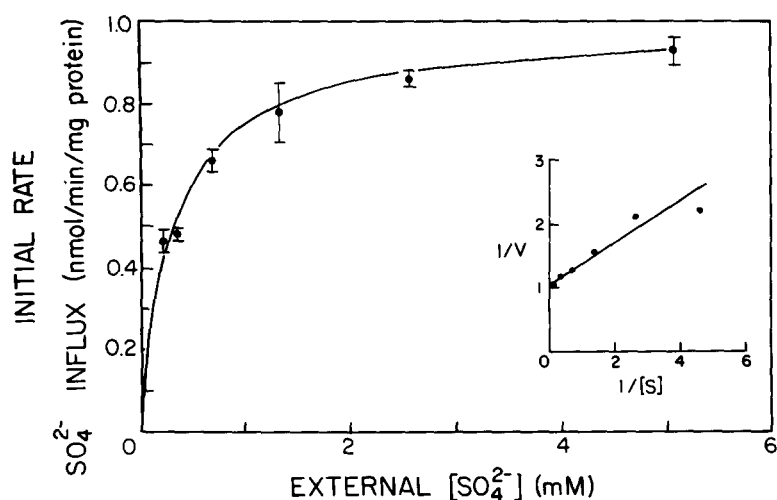


Fig. 2. Dependence of  $^{35}\text{SO}_4^{2-}$  influx on  $\text{SO}_4^{2-}$  concentration.  $[\text{SO}_4^{2-}]$  was varied by equimolar replacement with gluconate. Influx medium comprised 280 mM sucrose/10 mM  $\text{Na}^+$ /1.8 mM  $\text{Mg}^{2+}$ /10 mM Hepes, and the pH was adjusted to 7.3 with NaOH. Influx was initiated by replacing the preequilibrating HBHS with the labeled medium and was terminated at 2 min by rinsing three times with HBHS. Each data point is the mean of four determinations ( $\pm$  S.E.). Inset: double-reciprocal plot of the same data. The line was drawn using the slope and intercepts derived from the computer analysis of the data, according to the method of Eisenthal and Cornish-Bowden [12].

function of  $\text{SO}_4^{2-}$  concentration at three concentrations of  $\text{Cl}^-$  revealed that external  $\text{Cl}^-$  was a competitive inhibitor (Fig. 4). Computer analysis [14] of these data gave a  $K_i$  of  $15.5 \pm 1.4$  mM (mean  $\pm$  S.E.).

Since competitive inhibition by  $\text{Cl}^-$  suggested that  $\text{Cl}^-$  might be a substrate for the  $\text{SO}_4^{2-}$  exchanger, we examined the ability of  $\text{Cl}^-$  to trans-stimulate the efflux of  $\text{SO}_4^{2-}$  (Fig. 3C). External

NaCl (140 mM) stimulated initial  $^{35}\text{SO}_4^{2-}$  efflux by a factor of 3.3 over the control, which contained only sucrose and Hepes buffer. Thus, external  $\text{Cl}^-$  was able to trans-stimulate  $^{35}\text{SO}_4^{2-}$  efflux and appeared to be more effective than  $\text{SO}_4^{2-}$  itself.

We also examined the ability of internal  $\text{Cl}^-$  to trans-stimulate  $\text{SO}_4^{2-}$  influx. Since incubation in low- $\text{Cl}^-$  medium causes a decrease in steady-state

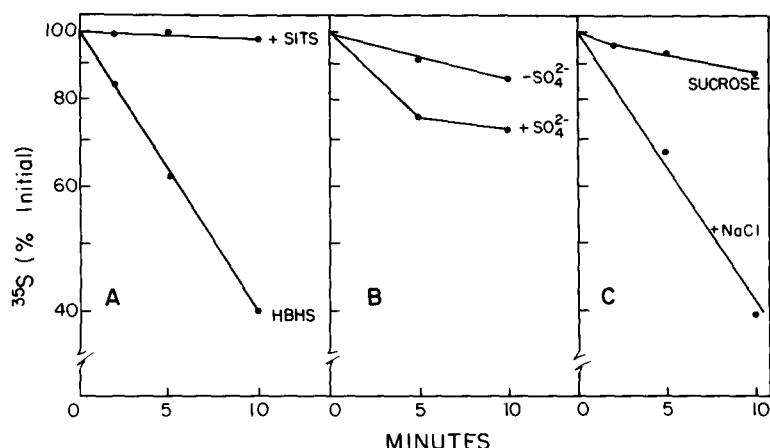


Fig. 3. Effect of SITS (A), external  $\text{SO}_4^{2-}$  (B) and external  $\text{Cl}^-$  (C) on efflux of  $^{35}\text{SO}_4^{2-}$ . Cells were loaded with  $^{35}\text{SO}_4^{2-}$  by incubation for 30–70 min in HBHS containing  $^{35}\text{SO}_4^{2-}$ . Efflux was initiated by replacing the labeled medium with 800 or 1000  $\mu\text{l}$  of the appropriate unlabeled medium and was terminated by rinsing three times with HBHS. Each point is the mean of four determinations. S.E. values are all less than 7% of the means. The ordinate is a logarithmic scale. (A) Effect of SITS. Efflux medium was HBHS, with or without 1 mM SITS. (B) Effect of external  $\text{SO}_4^{2-}$ . The  $\text{SO}_4^{2-}$ -containing efflux medium consisted of 10 mM  $\text{SO}_4^{2-}$ /20 mM  $\text{Na}^+$ /1 mM  $\text{K}^+$ /10 mM Hepes/270 mM sucrose. The  $\text{SO}_4^{2-}$ -free efflux medium comprised 1 mM  $\text{K}^+$ /10 mM Hepes/300 mM sucrose. (C) Effect of external  $\text{Cl}^-$ . The  $\text{Cl}^-$ -containing efflux medium consisted of 140 mM  $\text{Cl}^-$ /140 mM  $\text{Na}^+$ /10 mM Hepes. The  $\text{Cl}^-$ -free medium comprised 280 mM sucrose/10 mM Hepes. All media were adjusted to pH 7.3 with NaOH.

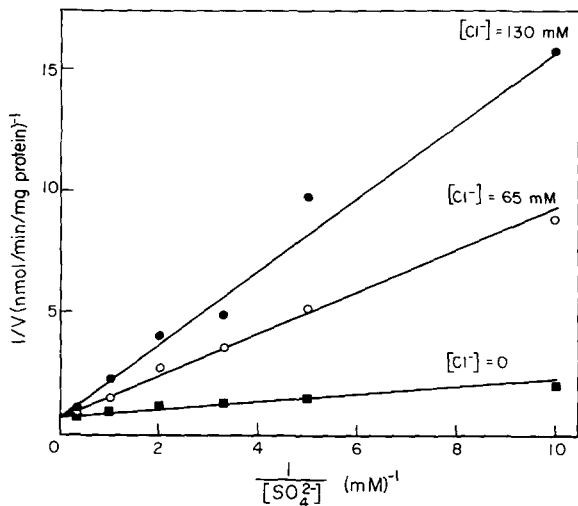


Fig. 4. Double-reciprocal plots of  $\text{Cl}^-$  inhibition of  $^{35}\text{SO}_4^{2-}$  influx. The  $\text{Cl}^-$ -free media comprised 4.1 mM  $\text{K}^+$ /0.71 mM  $\text{Mg}^{2+}$ /1.1 mM  $\text{Ca}^{2+}$ /11 mM glucose/7.7 mM gluconate/260 mM sucrose/10 mM Hepes. For the  $\text{Cl}^-$ -containing media, NaCl was added by equimolar reduction of sucrose. The concentration of  $\text{SO}_4^{2-}$  was varied by equimolar replacement of  $\text{SO}_4^{2-}$  with gluconate. Influx was initiated by replacing the preequilibrating HBHS with the  $^{35}\text{SO}_4^{2-}$ -containing media and was terminated at 2 min by rinsing three times with HBHS. Each point is the mean of two or three determinations. The lines were drawn using the slopes and intercepts derived from computer analysis of the data, according to the method of Cleland [14].

$\text{Cl}^-$  content of LRM55 cells [5], internal  $\text{Cl}^-$  was varied by preincubating the cells for 40 min in medium containing 7.3, 72, or 136 mM  $\text{Cl}^-$ .  $^{35}\text{SO}_4^{2-}$  uptake was then measured in medium containing 0.21 mM  $\text{SO}_4^{2-}$  and 7.3 mM  $\text{Cl}^-$ . The initial rates of uptake into cells preincubated in 7.3 and 72 mM external  $\text{Cl}^-$  were 47 and 75% of that into the control cells preincubated in 136 mM  $\text{Cl}^-$  (data not shown). Thus, internal  $\text{Cl}^-$  was able to trans-stimulate  $^{35}\text{SO}_4^{2-}$  influx.

#### Effect of chloride transport inhibitors on sulfate transport

Since our previous studies show that SITS, DIDS and furosemide inhibit  $\text{Cl}^-$  transport in LRM55 cells [5], we examined the effect of these compounds on  $\text{SO}_4^{2-}$  flux in these cells. SITS, a disulfonic acid stilbene derivative, was an effective inhibitor of  $\text{SO}_4^{2-}$  transport in LRM55 cells. At 1 mM, SITS inhibited  $\text{SO}_4^{2-}$  influx by 88% (Fig. 5A) and efflux by 94% (Fig. 3A).  $\text{SO}_4^{2-}$  influx was also inhibited 73% by 10  $\mu\text{M}$  DIDS, another stilbene derivative (results not shown). The diuretic furosemide was also an effective inhibitor; at 5 mM, furosemide produced 98% inhibition of  $\text{SO}_4^{2-}$  influx (Fig. 5B).

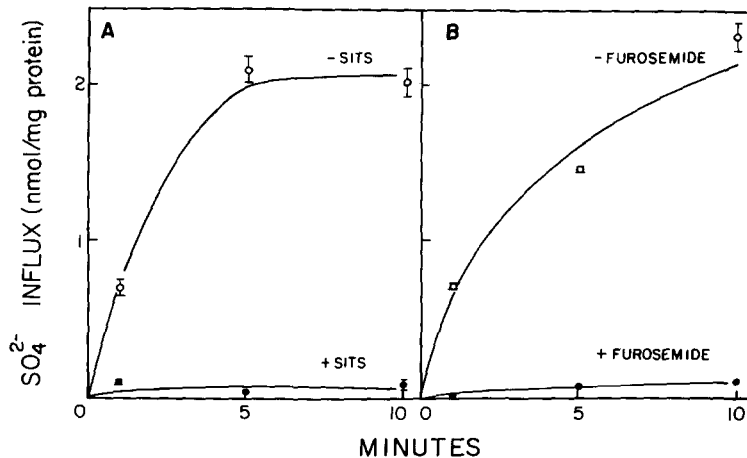


Fig. 5. Effect of SITS (A) and furosemide (B) on influx of  $^{35}\text{SO}_4^{2-}$ . Cells were preincubated for 30 min with (●) or without (○) the indicated inhibitor. Influx was initiated by replacing each preincubation medium with low- $\text{Cl}^-$  medium containing  $^{35}\text{SO}_4^{2-}$  and the inhibitor, and was terminated by rinsing three times with HBHS. Each point is the mean of three or four determinations ( $\pm$  S.E.). S.E. values are omitted where they are smaller than the symbols. (A) Effect of SITS. Influx medium was low- $\text{Cl}^-$  HBHS (○) or low- $\text{Cl}^-$  HBHS with 1 mM SITS (●). (B) Effect of furosemide. Influx media contained 18 mM Hepes, 19 mM glucose, 9.8 mM  $\text{Na}^+$ , 4.1 mM  $\text{K}^+$ , 6.6 mM  $\text{Cl}^-$ , 1.0 mM  $\text{Ca}^{2+}$ , 0.64 mM  $\text{Mg}^{2+}$ , 0.19 mM  $\text{SO}_4^{2-}$ , 1.2 mM phosphate, and either 5 mM furosemide (●) or 5 mM isethionate (○). All media were adjusted to pH 7.3 with NaOH.

### Effect of sulfate on chloride flux

Since  $\text{SO}_4^{2-}$  exchange transport was competitively inhibited by  $\text{Cl}^-$ , since its pharmacologic sensitivities were similar to those of the glial  $\text{Cl}^-/\text{HCO}_3^-$  exchanger, and since  $\text{Cl}^-$  acted as a substrate for the  $\text{SO}_4^{2-}$  porter, it appeared initially that  $\text{SO}_4^{2-}$  transport was carried out by the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger. If this hypothesis were correct, then external  $\text{SO}_4^{2-}$  ought to inhibit that portion of total  $\text{Cl}^-$  flux that is mediated by the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger. To test for such inhibition, we held external  $[\text{Cl}^-]$  constant at 35 mM and measured the rate of  $^{36}\text{Cl}^-$  uptake over a range of external  $\text{SO}_4^{2-}$  concentrations from 0.02 to 20 mM (Fig. 6). There was no significant inhibition of  $\text{Cl}^-$  influx at any concentration of  $\text{SO}_4^{2-}$ . Using the kinetic constants determined by the data shown in Fig. 4, we calculated that at 35 mM  $\text{Cl}^-$  and 20 mM  $\text{SO}_4^{2-}$ , the  $\text{Cl}^-$  exchanger would have been inhibited by 97% if  $\text{SO}_4^{2-}$  entered only on the major  $\text{Cl}^-/\text{HCO}_3^-$  exchanger. However, since  $\text{Cl}^-$  uptake via the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger is only 63% of total  $\text{Cl}^-$  uptake in LRM55 cells [5], 20 mM  $\text{SO}_4^{2-}$  ought to have caused 61% inhibition of total  $\text{Cl}^-$  influx. This experiment was repeated under several different, but analogous conditions, with no evidence of any ability of  $\text{SO}_4^{2-}$  to inhibit

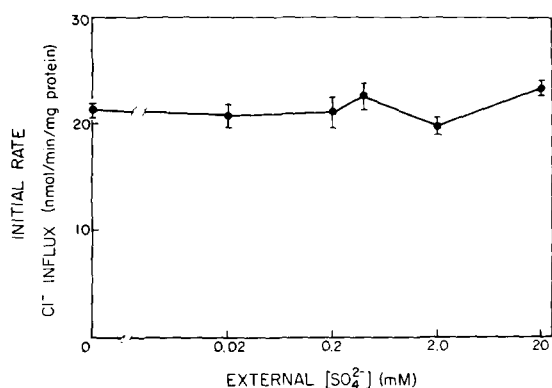


Fig. 6. Lack of effect of external  $\text{SO}_4^{2-}$  on influx of  $^{36}\text{Cl}^-$ . Influx was initiated by replacing the preequilibrating HBHS with influx media containing different concentrations of  $\text{SO}_4^{2-}$  and was terminated at 2 min by rinsing three times with HBHS.  $[\text{SO}_4^{2-}]$  was varied by the reciprocal equimolar variation of  $\text{Na}_2\text{SO}_4$  and sucrose. All influx media also contained 35 mM  $\text{Cl}^-$ , 4.1 mM  $\text{K}^+$ , 0.7 mM  $\text{Mg}^{2+}$  and 10 mM Hepes, with sucrose comprising the balance (total osmolality of 300 mM). Each point is the mean of four determinations ( $\pm$  S.E.).

$^{36}\text{Cl}^-$  uptake. The inability of  $\text{SO}_4^{2-}$  to inhibit  $\text{Cl}^-$  influx is not consistent with  $\text{SO}_4^{2-}$  transport by the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger.

We performed another series of experiments to examine this question further. If the  $\text{SO}_4^{2-}$  porter were the same porter as the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger as originally proposed, then consideration of the kinetic parameters for the two ions suggests that external  $\text{SO}_4^{2-}$  should be able to inhibit the transstimulation of  $\text{Cl}^-$  efflux by external  $\text{Cl}^-$ . That is, maximal  $\text{SO}_4^{2-}$  influx was only 1.2 nmol/min per

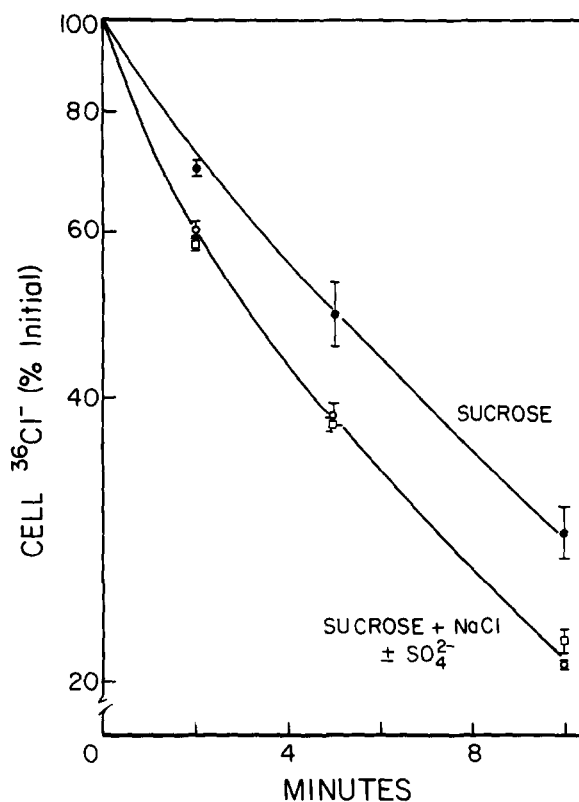


Fig. 7. Lack of inhibition of  $\text{Cl}^-$ -stimulated  $^{36}\text{Cl}^-$  efflux by external  $\text{SO}_4^{2-}$ . Cells were loaded with  $^{36}\text{Cl}^-$  by incubation for 60 min in HBHS containing  $^{36}\text{Cl}^-$ . Efflux was initiated by replacing the labeled HBHS with one of three unlabeled efflux media and was terminated by rinsing three times in HBHS. The efflux media contained: (●) 260 mM sucrose and 3 mM gluconate; (○) 130 mM NaCl and 3 mM gluconate; or (□) 130 mM NaCl and 3 mM  $\text{SO}_4^{2-}$ . In addition, all of these media also contained 4.1 mM  $\text{K}^+$ , 1.1 mM  $\text{Ca}^{2+}$ , 0.71 mM  $\text{Mg}^{2+}$ , 8 mM gluconate, 11 mM glucose and 10 mM Hepes. Each point is the mean of four determinations ( $\pm$  S.E.). Control experiments not presented here showed that gluconate itself did not significantly affect  $\text{Cl}^-$  efflux.

mg protein, which is very slow relative to  $\text{Cl}^-$  flux [5]. Therefore, if the two porters were identical, saturating concentrations of external  $\text{SO}_4^{2-}$  would be expected to retard the stimulating effect of external  $\text{Cl}^-$ . To test this, we examined the effect of external  $\text{SO}_4^{2-}$  on the stimulation of  $^{36}\text{Cl}^-$  efflux by external  $\text{Cl}^-$  (Fig. 7). In the absence of external  $\text{SO}_4^{2-}$ , external  $\text{Cl}^-$  alone (130 mM) stimulated efflux of  $^{36}\text{Cl}^-$  from the cells as expected (compare open circles and closed circles, Fig. 7). The addition of 3 mM  $\text{SO}_4^{2-}$  to the 130 mM  $\text{Cl}^-$  medium (open squares, Fig. 7) caused no change in the stimulating effect of external  $\text{Cl}^-$ . At these concentrations of  $\text{Cl}^-$  and  $\text{SO}_4^{2-}$ , if  $\text{SO}_4^{2-}$  were a substrate for the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger,  $\text{SO}_4^{2-}$  should have caused 56% inhibition of the  $\text{Cl}^-$ -stimulated portion of  $\text{Cl}^-$  efflux. This was not the case. In other similar experiments, the concentration of  $\text{SO}_4^{2-}$  was raised as high as 20 mM but there was still no inhibition of the  $\text{Cl}^-$ -stimulated efflux of  $^{36}\text{Cl}^-$ . Since  $\text{SO}_4^{2-}$  did not retard the stimulating effect of external  $\text{Cl}^-$ ,  $\text{SO}_4^{2-}$  does not appear to be a substrate for the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger.

#### Effect of potassium on sulfate flux

Like  $\text{SO}_4^{2-}$  transport,  $\text{K}^+/\text{Cl}^-$  cotransport in LRM55 cells is sensitive to furosemide [5]. It was thus important to determine whether  $\text{SO}_4^{2-}$  transport was mediated by the  $\text{K}^+/\text{Cl}^-$  cotransporter, although this seemed unlikely due to the inability of  $\text{SO}_4^{2-}$  to inhibit total  $\text{Cl}^-$  influx (Fig. 6). If  $\text{SO}_4^{2-}$  were cotransported with  $\text{K}^+$ ,  $\text{SO}_4^{2-}$  influx

should be dependent on external  $[\text{K}^+]$ . Therefore, we tested the effect of external  $[\text{K}^+]$  on  $^{35}\text{SO}_4^{2-}$  influx (Fig. 8A).  $\text{SO}_4^{2-}$  influx was the same whether external  $[\text{K}^+]$  was 4.1 mM or 102 mM. This contrasted sharply with the strong effect of  $\text{K}^+$  on  $\text{Cl}^-$  transport (Fig. 8B). These results suggest that the  $\text{K}^+/\text{Cl}^-$  porter is not a significant  $\text{SO}_4^{2-}$  porter.

#### Effects of other physiologic anions on sulfate influx

The results described above suggest that the  $\text{SO}_4^{2-}$  porter is a distinct porter which is able to carry  $\text{Cl}^-$  but is responsible for only a minor fraction of total  $\text{Cl}^-$  flux. It was thus of interest to determine whether there were other physiologically important substrates for this  $\text{SO}_4^{2-}$  carrier. We tested a series of organic anions for ability to inhibit  $\text{SO}_4^{2-}$  transport. These included nucleoside mono-, di- and triphosphates, cyclic AMP, glutamate, inorganic phosphate, pyridoxal phosphate and pyridoxamine phosphate. None of the compounds tested had any profound inhibitory effect on  $\text{SO}_4^{2-}$  influx (results not shown). Surprisingly, most of the test compounds stimulated rather than inhibited  $\text{SO}_4^{2-}$  influx. In contrast, the same compounds had no significant effect on  $\text{Cl}^-$  influx.

#### Discussion

Our results indicate that  $\text{SO}_4^{2-}$  transport in LRM55 cells occurs by an exchange mechanism. Although our kinetic data indicate that  $\text{Cl}^-$  is a substrate for the  $\text{SO}_4^{2-}$  exchanger, they also sug-

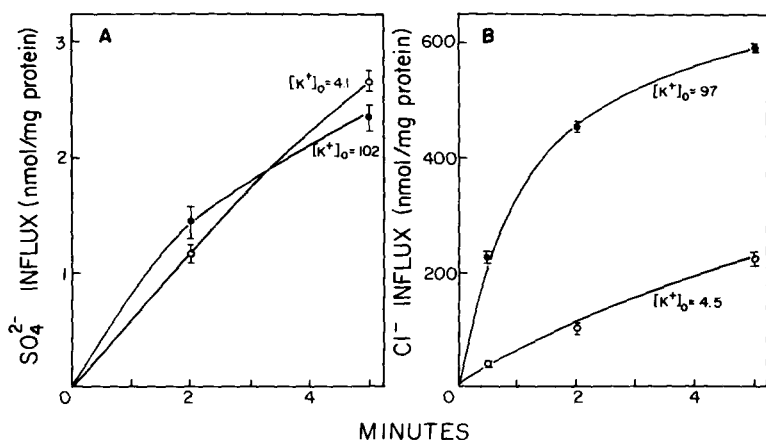


Fig. 8. Effect of external  $[\text{K}^+]$  on influx of  $^{35}\text{SO}_4^{2-}$  (A) and  $^{36}\text{Cl}^-$  (B). Cells were preincubated for 20 min in HBHS with  $[\text{K}^+]$  varied as indicated (in mM) by replacing  $\text{Na}^+$  with  $\text{K}^+$ . For (A)  $[\text{Cl}^-]$  was reduced to 7 mM by equiosmolar replacement of  $\text{NaCl}$  with sucrose. Influx was initiated by replacing each preincubation medium with medium containing  $^{35}\text{SO}_4^{2-}$  or  $^{36}\text{Cl}^-$  and was terminated by rinsing three times in ice-cold HBHS. Each point is the mean of three determinations ( $\pm$  S.E.); S.E. values are omitted where they are smaller than the symbols.

gest that the  $\text{SO}_4^{2-}$  transporter is not the same as the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger or the  $\text{K}^+/\text{Cl}^-$  cotransporter previously described in these cells [5]. The  $\text{SO}_4^{2-}$  transporter appears, instead, to be a different porter which is able to carry  $\text{Cl}^-$  but is responsible for only a minor fraction of total  $\text{Cl}^-$  flux. This conclusion is based on the failure of  $\text{SO}_4^{2-}$  to inhibit  $\text{Cl}^-$  flux. If  $\text{SO}_4^{2-}$  were transported on either of the two major anion carriers, it would inhibit  $\text{Cl}^-$  influx. Moreover, if  $\text{SO}_4^{2-}$  were transported on the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger, external  $\text{SO}_4^{2-}$  would inhibit  $\text{Cl}^-$ -stimulated  $\text{Cl}^-$  efflux. However, we observed no such inhibition. In addition,  $\text{K}^+$  did not stimulate  $\text{SO}_4^{2-}$  influx as it does  $\text{Cl}^-$  influx [5], providing further evidence that  $\text{Cl}^-$  does not enter by the  $\text{K}^+/\text{Cl}^-$  cotransporter.

The failure of  $\text{SO}_4^{2-}$  to inhibit  $\text{Cl}^-$  flux cannot be explained by application of Gunn's titratable carrier model [15] to the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger in LRM55 cells. This model assumes that the  $\text{Cl}^-/\text{HCO}_3^-$  carrier exists in two interconvertible states which are titratable with  $\text{H}^+$ : a doubly charged state which binds divalent  $\text{SO}_4^{2-}$  and a singly charged state which binds monovalent  $\text{Cl}^-$  or  $\text{HCO}_3^-$ . However, since an equilibrium exists between these two states, the binding of  $\text{SO}_4^{2-}$  to the doubly charged state would shift the equilibrium in the direction of the doubly charged state. As the  $\text{SO}_4^{2-}$  concentration is raised,  $\text{Cl}^-$  transport would be inhibited due to the effective removal of the singly charged state of the carrier. Since our experiments show that  $\text{SO}_4^{2-}$  does not inhibit  $\text{Cl}^-$  transport in LRM55 cells, application of the model of Gunn to the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger does not offer an explanation consistent with our results.

The kinetic data indicate that the  $\text{SO}_4^{2-}$  transporter in these cells can be saturated with  $\text{SO}_4^{2-}$ . The  $K_m$  for  $\text{SO}_4^{2-}$  (0.24 mM) is close to reported values for physiologic extracellular  $\text{SO}_4^{2-}$  [16]. Trans-stimulation of  $^{35}\text{SO}_4^{2-}$  efflux by external  $\text{SO}_4^{2-}$  is strong evidence for an exchange mechanism. Moreover, since  $\text{Cl}^-$  is a competitive inhibitor and since it trans-stimulates  $\text{SO}_4^{2-}$  transport from either side of the cell membrane,  $\text{Cl}^-$  appears to act as an alternative substrate for the  $\text{SO}_4^{2-}$  porter. The biphasic time-course of  $^{35}\text{SO}_4^{2-}$  influx is consistent with this interpretation. In the experiment shown in Fig. 1, influx was initiated by

exposing the cells to medium containing  $^{35}\text{SO}_4^{2-}$  and only 7 mM  $\text{Cl}^-$ . Since internal  $\text{Cl}^-$  is rapidly depleted when LRM55 cells are exposed to low- $\text{Cl}^-$  medium [5], and since  $\text{Cl}^-$  and  $\text{SO}_4^{2-}$  appear to be mutually exchangeable substrates of the  $\text{SO}_4^{2-}$  porter, one would expect an initial phase of concentrative  $\text{SO}_4^{2-}$  uptake as external  $\text{SO}_4^{2-}$  exchanges for  $\text{Cl}^-$  leaving the cell. After internal  $\text{Cl}^-$  becomes depleted over the course of the first 15 min of  $\text{SO}_4^{2-}$  uptake, the accumulated  $^{35}\text{SO}_4^{2-}$  exchanges with extracellular anions back across the membrane, reaching its steady-state level at 30 min.

The inhibitory effects of SITS, DIDS and furosemide suggest that the  $\text{SO}_4^{2-}$  porter may possess some structural similarity to the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger and the  $\text{K}^+/\text{Cl}^-$  porter, as well as anion porters in other types of cells. However, these compounds are strong inhibitors not only of anion exchange but also of other  $\text{Cl}^-$ -related processes in a wide variety of cell types [17–23]. Considering the wide reactivity of these compounds, it is not surprising that they inhibit the  $\text{SO}_4^{2-}$  exchanger along with other anion porters in LRM55 cells. It is noteworthy that these compounds are much more effective in inhibiting LRM55  $\text{SO}_4^{2-}$  flux than  $\text{Cl}^-$  flux [5].

The issue of a membrane carrier that is able to exchange a divalent anion for a monovalent anion has been discussed by several authors. As noted above, Gunn [15] proposed that such a carrier may exist in two interconvertible forms which are titratable with  $\text{H}^+$ . Jennings' conclusion [24] that  $\text{Cl}^-$  exchange for  $\text{SO}_4^{2-}$  is accompanied by movement of one  $\text{H}^+$  with each  $\text{SO}_4^{2-}$  supports Gunn's model in the erythrocyte anion exchanger. Callahan and Goldstein [25] suggested that carbonate crosses the human erythrocyte membrane as  $\text{NaCO}_3^-$ . By analogy,  $\text{NaSO}_4^-$  may be the sulfate species that is transported by an anion porter which seeks a monovalent substrate for electroneutral exchange. A mechanism of  $\text{SO}_4^{2-}$  influx accompanied by  $\text{Na}^+$  influx would unquestionably provide thermodynamic favorability for  $\text{SO}_4^{2-}$  uptake in our cells.

The physiologic role of the LRM55  $\text{SO}_4^{2-}$  exchange porter is not clear. This porter does not appear to provide a major pathway for  $\text{Cl}^-$ . It does not appear to carry glutamate, inorganic phosphate, or any of a number of phosphorylated



small organic molecules. Moreover, the strong inhibitory effect of  $\text{Cl}^-$  at physiologic concentrations of  $\text{Cl}^-$  and  $\text{SO}_4^{2-}$  suggests that even  $\text{SO}_4^{2-}$  transport by this porter is largely inhibited under physiologic conditions. Assuming extracellular physiologic concentrations of 130 mM for  $\text{Cl}^-$  and 0.3 mM for  $\text{SO}_4^{2-}$  [16], the velocity of  $\text{SO}_4^{2-}$  uptake in LRM55 cells would be no greater than 13% of  $V_{\max}$  for  $\text{SO}_4^{2-}$  under these conditions. However, even taking this  $\text{Cl}^-$  inhibition into account,  $\text{SO}_4^{2-}$  flux is quantitatively similar to the rates reported for astrocytic uptake of other compounds such as amino acids [10,23,26]. Moreover, quantitative cellular demand for  $\text{SO}_4^{2-}$  is probably not great in astrocytes. Sulfatides in the brain are found predominantly in myelin [27], although sulfatides and sulfated glycoproteins have also been identified in other brain membrane fractions [27,28]. Thus, LRM55  $\text{SO}_4^{2-}$  flux, small though it is in comparison to  $\text{Cl}^-$  flux, is very likely sufficient to meet cellular metabolic demands.

$\text{SO}_4^{2-}$  transport in LRM55 cells is in many respects similar to that in other types of cells. Like erythrocytes (reviewed in Refs. 6 and 7), LRM55 cells exchange  $\text{SO}_4^{2-}$  for  $\text{Cl}^-$ , but unlike erythrocytes, LRM55 cells do not transport  $\text{SO}_4^{2-}$  by the  $\text{Cl}^-/\text{HCO}_3^-$  exchanger. The similarities between anion transport in LRM55 cells and Ehrlich ascites tumor cells are very striking. In both cases,  $\text{SO}_4^{2-}$  is transported by homo-exchange or by hetero-exchange for  $\text{Cl}^-$  [8,9]. Also, SITS inhibits  $\text{SO}_4^{2-}$  flux almost completely but  $\text{Cl}^-$  flux only partially [5,9]. In both cases,  $\text{Cl}^-$  flux via the  $\text{SO}_4^{2-}$  pathway appears to account for only a minor fraction of total  $\text{Cl}^-$  flux [9,29]. Moreover, both types of cells appear to have two other mechanisms to account for the major portion of  $\text{Cl}^-$  flux: one carries out  $\text{Cl}^-$  self-exchange and does not transport  $\text{SO}_4^{2-}$ ; the other carries out DIDS-insensitive  $\text{K}^+/\text{Cl}^-$  cotransport [5,9,30–34].

Membrane transporters like these have been described in cells from many other tissues and species including, among others, rabbit kidney and ileum [35–38], rat liver and small intestine [39,40], Madin-Darby canine kidney cells [21] and human and sheep erythrocytes [41,42]. Considering the differences in cell type and function among all of these cells, it is remarkable that they are so similar with respect to their anion transport properties.

Many appear to have  $\text{SO}_4^{2-}$  exchange transporters which can also transport  $\text{Cl}^-$ . Many have  $\text{K}^+/\text{Cl}^-$  cotransporters and  $\text{Cl}^-/\text{HCO}_3^-$  exchangers. Although interest in each of these transporters initially focused on the specialized function of a particular type of cell, their widespread occurrence suggests that such mechanisms may not necessarily represent specialized functions of a particular cell type, but, like the ubiquitous  $(\text{Na}^+ + \text{K}^+)\text{-ATPase}$ , may be part of the general functional repertoire of mammalian cells.

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### References

- Gerschenfeld, H.M., Wald, F., Zadunaisky, J.A. and DeRobertis, E.D.P. (1959) *Neurology* 9, 412–425
- Kimelberg, H.K. and Bourke, R.S. (1982) in *Handbook of Neurochemistry* (Abel Lajtha, ed.), Vol. 1, pp. 31–67, Plenum Press, New York
- Kimelberg, H.K. (1981) *Biochim. Biophys. Acta* 646, 179–184
- Kimelberg, H.K. and Hirata, H. (1981) *Soc. Neurosci. Abstr.* 7, 698
- Wolpaw, E.W. and Martin, D.L. (1984) *Brain Res.* 297, 317–327
- Cabantchik, Z.I., Knauf, P.A. and Rothstein, A. (1978) *Biochim. Biophys. Acta* 515, 239–302
- Lowe, A.G. and Lambert, A. (1983) *Biochim. Biophys. Acta* 694, 353–374
- Villereal, M.L. and Levinson, C. (1976) *J. Cell. Physiol.* 89, 303–311
- Villereal, M.L. and Levinson, C. (1977) *J. Cell. Physiol.* 90, 553–564
- Martin, D.L. and Shain, W.G. (1979) *J. Biol. Chem.* 254, 7076–7084
- Morantz, R.A., Shain, W. and Cravioto, H. (1978) *J. Neurosurg.* 49, 84–92
- Eisenthal, R. and Cornish-Bowden, A. (1974) *Biochem. J.* 139, 715–720
- Wolpaw, E.W. (1984) Ph.D. Dissertation, University of Maryland, pp. 1–199
- Cleland, W.W. (1967) *Adv. Enzymol.* 29, 1–32
- Gunn, R.B. (1972) in *Oxygen Affinity of Hemoglobin and Red Cell Acid Base Status* (Astrup, P. and Rorth, M., eds.), pp. 823–827, Munksgaard, Copenhagen
- Van Harreveld, A., Ahmed, N. and Tanner, D.J. (1966) *Am. J. Physiol.* 210, 777–780

- 17 Brazy, P.C. and Gunn, R.B. (1976) *J. Gen. Physiol.* 68, 583–599
- 18 Brown, E.M., Pazoles, C.J., Creutz, C.E., Aurbach, G.D. and Pollard, H.B. (1978) *Proc. Natl. Acad. Sci. USA* 75, 876–880
- 19 Pazoles, C.J. and Pollard, H.B. (1978) *J. Biol. Chem.* 253, 3962–3969
- 20 Costa, T., Russell, L., Pert, C. and Rodbard, D. (1981) *Mol. Pharmacol.* 20, 470–476
- 21 McRoberts, J.A., Erlinger, S., Rindler, M.J. and Saier, M.H., Jr. (1982) *J. Biol. Chem.* 257, 2260–2266
- 22 Rindler, M.J., McRoberts, J.A. and Saier, M.H., Jr. (1982) *J. Biol. Chem.* 257, 2254–2259
- 23 Waniewski, R.A. and Martin, D.L. (1983) *Brain Res.* 268, 390–394
- 24 Jennings, M.L. (1976) *J. Membrane Biol.* 28, 187–205
- 25 Callahan, T.J. and Goldstein, D.A. (1978) *J. Gen. Physiol.* 72, 87–100
- 26 Martin, D.L. (1976) in *GABA in Nervous System Function* (Roberts, E., Chase, T.N. and Tower, D.B., eds.), pp. 347–386, Raven Press, New York
- 27 Harris, R.A. and Loh, H.H. (1979) *Res. Commun. Chem. Pathol. Pharmacol.* 24, 169–179
- 28 Simpson, D.L., Thorne, D.R. and Loh, H.H. (1976) *Biochemistry* 15, 5449–5457
- 29 Levinson, C. (1976) *J. Cell. Physiol.* 87, 235–244
- 30 Levinson, C. (1978) *J. Cell. Physiol.* 95, 23–32
- 31 Aull, F. (1980) *Biochim. Biophys. Acta* 599, 580–586
- 32 Geck, P., Pietrzyk, C., Burckhardt, B.C., Pfeiffer, B. and Heinz, E. (1980) *Biochim. Biophys. Acta* 600, 432–447
- 33 Aull, F. (1981) *Biochim. Biophys. Acta* 643, 339–345
- 34 Aull, F. (1982) *Biochim. Biophys. Acta* 688, 740–746
- 35 Brazy, P.C. and Dennis, V.W. (1981) *Am. J. Physiol.* 241, F300–F307
- 36 Kleinman, J.G., Ware, R.A. and Schwartz, J.H. (1981) *Biochim. Biophys. Acta* 648, 87–92
- 37 Langridge-Smith, J.E. and Field, M. (1981) *J. Membrane Biol.* 63, 207–214
- 38 Smith, P.L., Orellana, S.A. and Field, M. (1981) *J. Membrane Biol.* 63, 199–206
- 39 Cheng, S. and Levy, D. (1980) *J. Biol. Chem.* 255, 2637–2640
- 40 Liedtke, C.M. and Hopfer, U. (1982) *Am. J. Physiol.* 242, G272–G280
- 41 Dunham, P.B., Stewart, G.W. and Ellory, J.C. (1980) *Proc. Natl. Acad. Sci. USA* 77, 1711–1715
- 42 Dunham, P.B. and Ellory, J.C. (1981) *J. Physiol.* 318, 511–530